





Luminosity in a Line (e.g L α)

- L (erg/s) = transitions/sec * Energy of transition $-\Delta E = (E_2 - E_1) = hv = hc/\lambda$
 - transitions per second = N₂/τ₂
 where τ₂ = lifetime of excited state
 and N₂ = # of electrons in excited state

•
$$N_2/V = n_2 N_H/V = n_H$$

- $-N_2 = N_H(n_2/n_1) = n_H V (n_2/n_1)$ - so it reduces to finding (n_2/n_1)
- so it feduces to finding (n_2/n_1)
- in thermal equilibrium, this is easy (and so are a lot of other things...)



- $-J_v = B_v$ (Planck function)
- Boltzmann distribution of level populations
- every process strictly balance by its inverse ("Principal of Detailed Balance")
- "Einstein relations"

$$B_{21} = g_1/g_2 \ B_{12}$$
$$A_{21} = 2hv^3/c^2 \ B_{12}$$

- $C_{21} = g_1/g_2 \exp(\Delta E_{12}/kT) \quad C_{12}$ These depend only on atomic parameters, so they should be valid for any temperature!
- g₁ and g₂ are "statistical weights" (# of degenerate electron states in each level; e.g. 2n²)













- 2. Pressure (collisional) Broadening
- broadens levels; depends on...
 lifetimes of levels and frequency of collisions
- has lorentzian shape and characteristic width
 - $\phi^{\rm C}(\mathbf{v}) \, \mathrm{d}\mathbf{v} \sim \Delta^{\rm C} / [(\mathbf{v} \mathbf{v}_{\rm o})^2 + \Delta^{\rm C^2}] \, \mathrm{d}\mathbf{v}$

 $\Delta^{\rm C} \sim n\sigma (2kT/m)^{1/2}$

- "Damping" from classical H.O. formation
 - radiative damping v. collisional damping
 - broad wings
 - n term "explains" Luminosity Class in MK scheme

- 3. Doppler Broadening
- Maxwellian distribution of absorber/emitter (atom) velocities
- Lorentzian in atom frame, but Doppler shift $(\Delta\lambda/\lambda=v_r/c) \rightarrow gaussian profile$ $\phi^D(v) dv \sim 1/(\pi\Delta^D) e^{-[(v-v_o)/\Delta^D]^2} dv$

 $\Delta^{\rm D} = (v_{\rm T}/c) v_{\rm o} = v_{\rm o}/c \ (2kT/m)^{1/2}$

m=mass of atom!

- What else can cause Doppler broadening?
 - turbulence
 - rotation, flows, pulsation (not necessarily gaussian)





