







- bubble expands without exchanging energy with
- if density inside bubble drops off faster with height than density in surrounding
- $d \ln(P) / d \ln(T) > 2.5$  then stable against convection

## **Equations of State**

- need Equation of State
- e.g. "Ideal Gas"
  - PV=NkT P=(N/V)kT P=nkT
  - alternative form involving  $\rho = n * \underline{m}$
  - $-\underline{m}$  is average mass of a gas particle
- "mean molecular weight"  $\mu = \underline{m}/m_{H}$ 
  - $-\mu$  is temperature and composition dependent  $-P_r = \rho_r k T_r / \mu_r m_H$
  - relates pressure with Temperature and composition at each radius

- Mass Fractions rather than numbers...
  X=M<sub>H</sub>/M Y=M<sub>He</sub>/M Z=M<sub>metal</sub>/M X+Y+Z=1
  e.g. neutral 1/μ = X+(1/4)Y+<1/A>Z
  solar: X=.70, Y=.28, Z=.02, A~15.5 so μ~1.3
  e.g. 100% ionized 1/μ = 2X+(3/4)Y+(1/2)Z
  solar: X=.70, Y=.28, Z=.02, A~15.5 so μ~0.62
- Radiation Pressure  $P_{rad} = (1/3)aT^4$
- Total Pressure =  $P_{gas} + P_{rad} + ...$

